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Like the author's other companion books, The Chemistry Companion provides high quality information in unique one-page-per-topic presentations that do not overburden and distract with excessive details. The book offers concise summaries of general chemistry concepts, easily accessible in a convenient, reader-friendly format. Suitable as an introductory Volume 68 in the internationally acclaimed Advances in Clinical Chemistry contains chapters authored by world renowned clinical laboratory scientists, physicians and research scientists. The serial provides the latest and most up-to-date technologies related to the field of Clinical Chemistry and is the benchmark for novel analytical approaches in the clinical laboratory. Expertise of international contributors Latest cutting-edge technologies Comprehensive in scope Synthetic Inorganic Chemistry: New Perspectives presents summaries of the work of some of the most creative researchers in the field. The book highlights the most novel approaches and burgeoning applications of synthetic inorganic chemistry in development. Topics include non-precious metals in catalysis, smart inorganic polymers, new inorganic therapeutics, new photocatalysts for hydrogen production, and more. As the first volume in the Developments in Inorganic Chemistry series, this work is a valuable resource for students and researchers working in inorganic chemistry and material science. Illustrates the scope and vitality of modern synthetic inorganic chemistry Shows the centrality of inorganic chemistry, addressing a variety of global challenges Serves to define the current, important and expanding roles of synthetic inorganic chemistry in interdisciplinary areas such as materials science, synthetic organic chemistry, homogeneous and heterogeneous catalysis A Q&A Approach to Organic Chemistry is a book of leading questions that begins with atomic orbitals and bonding. All critical topics are covered, including bonding,

nomenclature, stereochemistry, conformations, acids and bases, oxidations, reductions, substitution, elimination, acyl addition, acyl substitution, enolate anion reactions, the Diels-Alder reaction and sigmatropic rearrangements, aromatic chemistry, spectroscopy, amino acids and proteins, and carbohydrates and nucleosides. All major reactions are covered. Each chapter includes end-of-chapter homework questions with the answer keys in an Appendix at the end of the book. This book is envisioned to be a supplementary guide to be used with virtually any available undergraduate organic chemistry textbook. This book allows for a "self-guided" approach that is useful as one studies for a coursework exam or as one reviews organic chemistry for postgraduate exams. Key Features: Allows a "self-guided tour" of organic chemistry Discusses all important areas and fundamental reactions of organic chemistry Classroom tested Useful as a study guide that will supplement most organic chemistry textbooks Assists one in study for coursework exams or allows one to review organic chemistry for postgraduate exams Includes 21 chapters of leading questions that covers all major topics and major reactions of organic chemistry Physical Chemistry: Concepts and Theory provides a comprehensive overview of physical and theoretical chemistry while focusing on the basic principles that unite the sub-disciplines of the field. With an emphasis on multidisciplinary, as well as interdisciplinary applications, the book extensively reviews fundamental principles and presents recent research to help the reader make logical connections between the theory and application of physical chemistry concepts. Also available from the author: Physical Chemistry: Multidisciplinary Applications (ISBN 9780128005132). Describes how materials behave and chemical reactions occur at the molecular and atomic levels Uses theoretical constructs and mathematical computations to explain chemical properties and describe behavior of molecular and condensed matter Demonstrates the connection between math and chemistry and how to use math as a powerful tool to predict the properties of chemicals Emphasizes the intersection of chemistry, math, and physics and the resulting applications across many disciplines of science Comprehensive Natural Products Chemistry ALERT: Before you purchase, check with your instructor or review your course syllabus to ensure that you select the correct ISBN. Several versions of Pearson's MyLab & Mastering products exist for each title, including customized versions for individual schools, and registrations are not transferable. In addition, you may need a CourseID, provided by your instructor, to register for and use Pearson's MyLab & Mastering products. Packages Access codes for Pearson's MyLab & Mastering products may not be included when purchasing or renting from companies other than Pearson; check with the seller before completing your purchase. Used or rental books If you rent or purchase a used book with an access code, the access code may have been redeemed previously and you may have to purchase a new access code. Access codes Access codes that are purchased from sellers other than Pearson carry a higher risk of being either the wrong ISBN or a previously redeemed code. Check with the seller prior to purchase. xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx For two-semester general chemistry courses Bestselling author Niva Tro has always believed "the behavior of matter is determined by the properties of molecules and atoms" to be the most

important discovery in scientific knowledge. This idea is the entire factor for his seminal new text-- Chemistry: Structure and Properties. Dr. Tro emphasizes the relationship between structure and properties, establishes a unique approach to teaching chemistry by presenting atomic and bonding theories early in the text, and stresses key themes throughout. The book is organized to present chemistry as a logical, cohesive story from the microscopic to the macroscopic, so students can fully grasp the theories and framework behind the chemical facts. Every topic has been carefully crafted to convey to students that the relationship between structure and properties is the thread that weaves all of chemistry together. While developed independently of other Tro texts, Chemistry: Structure and Properties incorporates the author's vivid writing style, chemical rigor, dynamic multi-level images, and tested features. His consistent conceptual focus and step-by-step problem-solving framework encourages you to think through processes rather than simply memorize content. Interactive media within MasteringChemistry® complements the book's problem-solving approach, thus creating a comprehensive program that enables you to learn both in and out of the classroom. This program presents a better teaching and learning experience for you. Personalized learning with MasteringChemistry: This online homework, tutorial, and assessment program is designed to improve results by helping you quickly master concepts. You'll benefit from self-paced tutorials, featuring specific wrong-answer feedback and hints that emulate the office-hour experience. Developed with a central theme and by a teaching community: As part of a community that teaches with the understanding that matter is composed of particles and the structure of those particles determines the properties of matter, Dr. Tro took great lengths in the text to ensure that everything from organization, art, and pedagogy reinforce this theme. The result of this emphasis is that the topic order has been constructed to make key connections earlier, stronger, and more often than the traditional approach. Linking conceptual understanding with problem-solving skills: Throughout each chapter, numerous Conceptual Connections encourage comprehension of the most complex concepts while a consistent step-by-step framework in the worked examples allows you to think logically through the problem-solving process. Visualizing and understanding chemistry: Revolutionary multipart images illustrate and reinforce the theme of the text and allows you to see and experience the molecules responsible for the structures and properties of matter. Note: You are purchasing a standalone product; MasteringChemistry does not come packaged with this content. If you would like to purchase both the physical text and MasteringChemistry search for ISBN-10: 0321729730/ISBN-13: 9780321729736. That package includes ISBN-10: 0321834682/ISBN-13: 9780321834683 and ISBN-10: 0321934105/ISBN-13: 9780321934109. MasteringChemistry is not a self-paced technology and should only be purchased when required by an instructor. Supramolecular chemistry is 'chemistry beyond the molecule' - the chemistry of molecular assemblies and intermolecular bonds. It is one of today's fastest growing disciplines, crossing a range of subjects from biological chemistry to materials science; and from synthesis to spectroscopy. Supramolecular Chemistry is an up-to-date, integrated textbook that tells the

newcomer to the field everything they need to know to get started. Assuming little in the way of prior knowledge, the book covers the concepts behind the subject, its breadth, applications and the latest contemporary thinking in the area. It also includes coverage of the more important experimental and instrumental techniques needed by supramolecular chemists. The book has been thoroughly updated for this second edition. In addition to the strengths of the very popular first edition, this comprehensive new version expands coverage into a broad range of emerging areas. Clear explanations of both fundamental and nascent concepts are supplemented by up-to-date coverage of exciting emerging trends in the literature. Numerous examples and problems are included throughout the book. A system of "key references" allows rapid access to the secondary literature, and of course comprehensive primary literature citations are provided. A selection of the topics covered is listed below. Cation, anion, ion-pair and molecular host-guest chemistry Crystal engineering Topological entanglement Clathrates Self-assembly Molecular devices Dendrimers Supramolecular polymers Microfabrication Nanoparticles Chemical emergence Metal-organic frameworks Gels Ionic liquids Supramolecular catalysis Molecular electronics Polymorphism Gas sorption Anion-pinteractions Nanochemistry

Supramolecular Chemistry is a must for both students new to the field and for experienced researchers wanting to explore the origins and wider context of their work. Review: "At just under 1000 pages, the second edition of Steed and Atwood's Supramolecular Chemistry is the most comprehensive overview of the area available in textbook form...highly recommended." —Chemistry World, August 2009

Joel Karty has dedicated nearly a decade developing a teaching approach and textbook that is organized by mechanism, promotes learning by doing, and provides students with the background and support they need to be successful in organic chemistry as well as pre-professional placement exams like the MCAT. Karty's organization, conversational writing style, and interactive pedagogy facilitate understanding rather than memorization and place the emphasis back on mechanisms." This print companion to MindTap General Chemistry: Atoms First presents the narrative, figures, tables and example problems—but no graded problems or assessments. Students must use MindTap to complete the interactive activities, exercises, and assignments. The atoms first organization introduces students to atoms and molecules earlier and delays math-intensive problem-solving to later in the semester. This gives students a stronger conceptual framework to help them succeed in the course. In addition, the narrative provides greater emphasis on the historical development of the atomic nature of matter and atomic structure. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version. This book provides an introduction to physical chemistry that is directed toward applications to the biological sciences. Advanced mathematics is not required. This book can be used for either a one semester or two semester course, and as a reference volume by students and faculty in the biological sciences. Chapter 1. The Vine -- Chapter 2. Composition of Grape Must -- Chapter 3. Must Aromas -- Chapter 4. Composition of Wine -- Chapter 5. Polyphenols -- Chapter 6. Sugars: Structure and Classification --

Chapter 7. Sugars in Must -- Chapter 8. Carboxylic Acids: Structure and Properties -- Chapter 9. Grape Acids -- Chapter 10. The Relationship between Must Composition and Quality -- Chapter 11. The Transformation of Must Into Wine -- Chapter 12. Nitrogen Compounds -- Chapter 13. Acid-Base Equilibria in Wine -- Chapter 14. Buffering Capacity of Wines -- Chapter 15. Precipitation Equilibria in Wine -- Chapter 16. Changes in Acidity After Fermentation -- Chapter 17. Redox phenomena in Must and Wine -- Chapter 18. The Colloidal State -- Chapter 19. Wine Colloids -- Chapter 20. Inorganic Material and Metal Casse -- Chapter 21. Chemical Aging -- Chapter 22. Aging -- Chapter 23. Biological Aging.

Physical Chemistry for the Biosciences has been optimized for a one-semester introductory course in physical chemistry for students of biosciences. This textbook provides essential information for students of inorganic chemistry or for chemists pursuing self-study. The presentation of topics is made with an effort to be clear and concise so that the book is portable and user friendly. Inorganic Chemistry 2E is divided into five major themes (structure, condensed phases, solution chemistry, main group and coordination compounds) with several chapters in each. There is a logical progression from atomic structure to molecular structure to properties of substances based on molecular structures, to behavior of solids, etc. The author emphasizes fundamental principles-including molecular structure, acid-base chemistry, coordination chemistry, ligand field theory, and solid state chemistry -and presents topics in a clear, concise manner. There is a reinforcement of basic principles throughout the book. For example, the hard-soft interaction principle is used to explain hydrogen bond strengths, strengths of acids and bases, stability of coordination compounds, etc. The book contains a balance of topics in theoretical and descriptive chemistry. New to this Edition: New and improved illustrations including symmetry and 3D molecular orbital representations Expanded coverage of spectroscopy, instrumental techniques, organometallic and bio-inorganic chemistry More in-text worked-out examples to encourage active learning and to prepare students for their exams • Concise coverage maximizes student understanding and minimizes the inclusion of details students are unlikely to use. • Discussion of elements begins with survey chapters focused on the main groups, while later chapters cover the elements in greater detail. • Each chapter opens with narrative introductions and includes figures, tables, and end-of-chapter problem sets. This text contains detailed worked solutions to all the end-of-chapter exercises in the textbook Organic Chemistry. Notes in tinted boxes in the page margins highlight important principles and comments. Visualizing Everyday Chemistry is for a one-semester course dedicated to introducing chemistry to non-science students. It shows what chemistry is and what it does, by integrating words with powerful and compelling visuals and learning aids. With this approach, students not only learn the basic principles of chemistry but see how chemistry impacts their lives and society. The goal of Visualizing Everyday Chemistry is to show students that chemistry is important and relevant, not because we say it is but because they see it is. Introductory Chemistry creates light bulb moments for students and provides unrivaled support for instructors! Highly visual, interactive multimedia tools are an extension of Kevin Revell's distinct author voice and help students develop

critical problem solving skills and master foundational chemistry concepts necessary for success in chemistry. Essentials of Computational Chemistry provides a balanced introduction to this dynamic subject. Suitable for both experimentalists and theorists, a wide range of samples and applications are included drawn from all key areas. The book carefully leads the reader thorough the necessary equations providing information explanations and reasoning where necessary and firmly placing each equation in context. First Published in 1999. Routledge is an imprint of Taylor & Francis, an informa company. Natural products play an integral and ongoing role in promoting numerous aspects of scientific advancement, and many aspects of basic research programs are intimately related to natural products. With articles written by leading authorities in their respective fields of research, Studies in Natural Products Chemistry, Volume 37 presents current frontiers and future guidelines for research based on important discoveries made in the field of bioactive natural products. It is a valuable source for researchers and engineers working in natural products and medicinal chemistry. Describes the chemistry of bioactive natural products Contains contributions by leading authorities in the field A valuable source for researchers and engineers working in natural product and medicinal chemistry This comprehensive handbook presents fundamental aspects, fabrication techniques, introductory materials on microbiology and chemistry, measurement techniques, and applications of microfluidics and nanofluidics. The second volume focuses on topics related to experimental and numerical methods. It also covers fabrication and applications in a variety of areas, from aerospace to biological systems. Reflecting the inherent nature of microfluidics and nanofluidics, the book includes as much interdisciplinary knowledge as possible. It provides the fundamental science background for newcomers and advanced techniques and concepts for experienced researchers and professionals. In Organic Chemistry, 3rd Edition, Dr. David Klein builds on the phenomenal success of the first two editions, which presented his unique skills-based approach to learning organic chemistry. Dr. Klein's skills-based approach includes all of the concepts typically covered in an organic chemistry textbook, and places special emphasis on skills development to support these concepts. This emphasis on skills development in unique SkillBuilder examples provides extensive opportunities for two-semester Organic Chemistry students to develop proficiency in the key skills necessary to succeed in organic chemistry. 'Environmental Chemistry' offers an accessible introduction to the chemical principles and concepts, and applies them to relevant environmental situations and issues. CliffsQuickReview course guides cover the essentials of your toughest classes. You're sure to get a firm grip on core concepts and key material and be ready for the test with this guide at your side. Whether you're new to biosynthesis, integrated metabolism, and the molecular cloning of DNA or just brushing up on those topics, CliffsQuickReview Organic Chemistry II can help. This guide introduces each topic, defines key terms, and walks you through each sample problem step-by-step. In no time, you'll be ready to tackle other concepts in this book such as The reactions of aromatic compounds Aldol, cross-aldol, and ketonic aldol condensation Ester formation Oxidation Mass spectra CliffsQuickReview Organic Chemistry II acts as

a supplement to your textbook and to classroom lectures. Use this reference in any way that fits your personal style for study and review — you decide what works best with your needs. You can either read the book from cover to cover or just look for the information you want and put it back on the shelf for later. What's more, you can Get a glimpse of what you'll gain from a chapter by reading through the introduction at the beginning of each chapter Study your formulas in the Summary of Preparations or the Summary of Reactions Use the glossary to find key terms fast. With titles available for all the most popular high school and college courses, CliffsQuickReview guides are a comprehensive resource that can help you get the best possible grades. "Chemistry is designed for the two-semester general chemistry course. For many students, this course provides the foundation to a career in chemistry, while for others, this may be their only college-level science course. As such, this textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them. The text has been developed to meet the scope and sequence of most general chemistry courses. At the same time, the book includes a number of innovative features designed to enhance student learning. A strength of Chemistry is that instructors can customize the book, adapting it to the approach that works best in their classroom."--Openstax College website. Gain a clear understanding of pathophysiology and lab testing! Clinical Chemistry: Fundamentals and Laboratory Techniques prepares you for success as a medical lab technician by simplifying complex chemistry concepts and lab essentials including immunoassays, molecular diagnostics, and quality control. A pathophysiologic approach covers diseases that are commonly diagnosed through chemical tests — broken down by body system and category — such as respiratory, gastrointestinal, and cardiovascular conditions. Written by clinical chemistry educator Donna Larson and a team of expert contributors, this full-color book is ideal for readers who may have minimal knowledge of chemistry and are learning laboratory science for the first time. Full-color illustrations and design simplify complex concepts and make learning easier by highlighting important material. Case studies help you apply information to real-life scenarios. Pathophysiology and Analytes section includes information related to diseases or conditions, such as a biochemistry review, disease mechanisms, clinical correlation, and laboratory analytes and assays. Evolve companion website includes case studies and animations that reinforce what you've learned from the book. Laboratory Principles section covers safety, quality assurance, and other fundamentals of laboratory techniques. Review questions at the end of each chapter are tied to the learning objectives, helping you review and retain the material. Critical thinking questions and discussion questions help you think about and apply key points and concepts. Other Aspects of Clinical Chemistry section covers therapeutic drug monitoring, toxicology, transplantation, and emergency preparedness. Learning objectives in each chapter help you to remember key points or to analyze and synthesize concepts in clinical chemistry. A list of key words is provided at the beginning of each chapter, and these are also bolded in the text. Chapter summaries consist of bulleted lists and tables highlighting the

most important points of each chapter. A glossary at the back of the book provides a quick reference to definitions of all clinical chemistry terms. "Can Munowitz write or what!" exclaimed one advance reviewer of this extraordinary new text. Class-tested and thoughtfully designed for student engagement, *Principles of Organic Chemistry* provides the tools and foundations needed by students in a short course or one-semester class on the subject. This book does not dilute the material or rely on rote memorization. Rather, it focuses on the underlying principles in order to make accessible the science that underpins so much of our day-to-day lives, as well as present further study and practice in medical and scientific fields. This book provides context and structure for learning the fundamental principles of organic chemistry, enabling the reader to proceed from simple to complex examples in a systematic and logical way. Utilizing clear and consistently colored figures, *Principles of Organic Chemistry* begins by exploring the step-by-step processes (or mechanisms) by which reactions occur to create molecular structures. It then describes some of the many ways these reactions make new compounds, examined by functional groups and corresponding common reaction mechanisms. Throughout, this book includes biochemical and pharmaceutical examples with varying degrees of difficulty, with worked answers and without, as well as advanced topics in later chapters for optional coverage. Incorporates valuable and engaging applications of the content to biological and industrial uses Includes a wealth of useful figures and problems to support reader comprehension and study Provides a high quality chapter on stereochemistry as well as advanced topics such as synthetic polymers and spectroscopy for class customization Medicinal Chemistry: Fundamentals presents the cycle of the life of drugs, their physico-chemical properties, and consequences that arise in development. The fundamental concepts of Medicinal Chemistry (pharmacophore, prodrugs, Lipinsky rules) are also presented, including discussions on specific concerns of the European Pharmacopeia - the industrialist's bible - its role, and a description of the monographs of active principles. Defines the lifecycle of drugs Explains the physico-chemical properties and consequences of a drug Studies the fundamental concepts of medicinal chemistry Describes the active ingredient monographs Analytical Chemistry, Second Edition covers the fundamental principles of analytical chemistry. This edition is organized into 30 chapters that present various analytical chemistry methods. This book begins with a core of six chapters discussing the concepts basic to all of analytical chemistry. The fundamentals, concepts, applications, calculations, instrumentation, and chemical reactions of five major areas of analytical chemistry, namely, neutralization, potentiometry, spectroscopy, chromatography, and electrolysis methods, are emphasized in separate chapters. Other chapters are devoted to a discussion of precipitation and complexes in analytical chemistry. Principles and applications and the relationship of these reactions to the other areas are stressed. The remaining chapters of this edition are devoted to the laboratory. A chapter discusses the basic laboratory operations, with an emphasis on safety. This topic is followed by a series of experiments designed to reinforce the concepts developed in the chapters. This book is designed for introductory courses in

analytical chemistry, especially those shorter courses servicing chemistry majors and life and health science majors. Helping you focus on mastering the quantitative skills and conceptual knowledge you need to get a true understanding of chemistry, this text continues the tradition of relevance that makes it so effective. Now including MasteringChemistry, the online homework, tutorial, and assessment product with a demonstrated record of helping students quickly master concepts, this edition includes new opportunities for you to practice key concepts. MasteringChemistry provides seamless synergy with the text to create a dynamic learning program that enables you to learn both in and out of the classroom. A Concise Text-Book of Organic Chemistry is a handy guide for chemistry students preparing for Advanced Level certificates. The nature of organic chemistry, compared with that of inorganic chemistry, is basically the chemistry of carbon. The book focuses on the arrangements and changes of the atoms inside the carbon molecules. The molecular formulas of organic compounds are therefore studied, including alkanes and their derivatives known as aliphatic or fatty acids, as well as the hydrocarbons of the benzene series and derivatives known as the aromatic compounds. The aliphatic amines as derivatives of ammonia resulting from the substitution of the hydrogen atoms by alkyl groups are described. The formula for methane, although at present is convenient for general purposes, is shown to be not a true representative of the actual arrangement in which four H radicals are grouped around the carbon atom. Castor oil, linseed, and other drying oils are also examined in terms of their glyceride (of other long chain unsaturated acids) content. Carbohydrates, divided as monosaccharides, polysaccharides, and glycosides, are discussed as to their empirical composition. The several methods and reagents for synthesizing organic compounds are explained, using the simple aliphatic organic compounds as an example. The aromatic series of organic compounds, such as the benzene series of hydrocarbons, and the aromatic sulfonic acids, phenols, and ethers are then analyzed. This book is suitable for students of organic chemistry and for those preparing for tests in the General Certificate of Education and for the Ordinary National Certificate. Readers related to agricultural, medical, pharmaceutical, and technological and technical courses can find this guide relevant. Green Chemistry: An Inclusive Approach provides a broad overview of green chemistry for researchers from either an environmental science or chemistry background, starting at a more elementary level, incorporating more advanced concepts, and including more chemistry as the book progresses. Every chapter includes recent, state-of-the-art references, in particular, review articles, to introduce researchers to this field of interest and provide them with information that can be easily built upon. By bringing together experts in multiple subdisciplines of green chemistry, the editors have curated a single central resource for an introduction to the discipline as a whole. Topics include a broad array of research fields, including the chemistry of Earth's atmosphere, water and soil, the synthesis of fine chemicals, and sections on pharmaceuticals, plastics, energy related issues (energy storage, fuel cells, solar, and wind energy conversion etc., greenhouse gases and their handling, chemical toxicology issues of everyday products (from perfumes to detergents or clothing), and

environmental policy issues. Introduces the topic of green chemistry with an overview of key concepts Expands upon presented concepts with the latest research and applications, providing both the breadth and depth researchers need Includes a broad range of application based problems to make the content accessible for professional researchers and undergraduate and graduate students Authored by experts in a broad range of fields, providing insider information on the aspects or challenges of a given field that are most important and urgent

"Atoms First seems to be the flavor of the year in chemistry textbooks, but many of them seem to be little more than rearrangement of the chapters. It takes a master like McQuarrie to go back to the drawing board and create a logical development from smallest to largest that makes sense to students."---Hal Harris, University of Missouri-St. Louis "McQuarrie's book is extremely well written, the order of topics is logical, and it does a great job with both introductory material and more advanced concepts. Students of all skill levels will be able to learn from this book."---Mark Kearley, Florida State University

This new fourth edition of General Chemistry takes an atoms-first approach from beginning to end. In the tradition of McQuarrie's many previous works, it promises to be another groundbreaking text. This superb new book combines the clear writing and wonderful problems that have made McQuarrie famous among chemistry professors and students worldwide. Presented in an elegant design with all-new illustrations, it is available in a soft-cover edition to offer professors a fresh choice at an outstanding value. Student supplements include an online series of descriptive chemistry Interchapters, a Student Solutions Manual, and an optional state-of-the-art Online Homework program. For adopting professors, an Instructor's Manual and a CD of the art are also available. MindTap General Chemistry is a personalized teaching and learning experience that allows instructors to control what students see and focus on relevant assignments that guide them to analyze, apply, and improve thinking. Seamlessly integrating simulations, videos and diagnostic quizzes, it helps students achieve course learning outcomes by bringing chemistry to life. Measure skills and outcomes with ease using powerful analytics that provide a visual dashboard with at-a-glance performance and engagement data that is used to provide direction regarding class and student needs. This version is accompanied by a print text that includes the narrative from the MindTap General Chemistry course.

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